SOLUTIONS

THERMODYNAMICS PRACTICE PROBLEMS FOR NON-TECHNICAL MAJORS

Thermodynamic Properties

1. If an object has a weight of 10 lbf on the moon, what would the same object weigh on Jupiter?

$g_{Jupiter} = 75\,\frac{ft}{sec^2}$ $\quad g_{Moon} = 5.4\,\frac{ft}{sec^2}$ $\quad g_c = 32\,\frac{lbm\text{-}ft}{lbf\text{-}sec^2}$

$$|W| = \frac{mg}{g_c} \Rightarrow m = \frac{|W|_{moon}\, g_c}{g_{moon}} = \frac{10 \times 32}{5.4} = 59.26 \text{ lbm}$$

$$|W|_{Jupiter} = \frac{m g_{Jupiter}}{g_c} = \frac{59.26 \times 75}{32} = \boxed{139 \text{ lbf}}$$

2. An object that weighs 50 lbf on earth is moved to Saturn where its new weight is 105 lbf. What is the acceleration due to gravity on Saturn?

$g_{Earth} = 32\,\frac{ft}{sec^2}$ $\quad g_c = 32\,\frac{lbm\text{-}ft}{lbf\text{-}sec^2}$

50 lbf on earth ⇒ 50 lbm

$$|W| = \frac{mg}{g_c} \Rightarrow g = \frac{|W|\, g_c}{m} = \frac{105 \times 32}{50} = \boxed{67.2\,\frac{ft}{sec^2}}$$

3. Define, using equations, specific volume (v) and density (ρ). What is the mathematical relationship between these two terms?

$$v = \frac{V}{m}, \quad \rho = \frac{m}{V}, \quad v = \frac{1}{\rho} \text{ or } \rho = \frac{1}{v}$$

Temperature and Pressure Measurements

4. (a) Define temperature.
   (b) What is the absolute temperature scale corresponding to Fahrenheit?
   (c) Convert 100° F to that absolute scale.

(a) Temperature: a measure of molecular activity of a substance.

(b) Rankine

(c) °R = °F + 460 ⇒ 100° F converts to 560° R

Problems And Solutions In Thermodynamics

Rachel S Tattersall

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