

SENIOR INTER CHEMISTRY

Important Questions

1. SOLID STATE (4 Marks)

Short Answer type Questions

1. Derive Bragg's equation.
2. Calculate the efficiency of packing in case of a metal of body - centered cubic crystal.

(Note: Also learn F.C.C.)

3. Describe the two main types of semi-conductors and contrast their conduction mechanism.

Very Short Answer type Questions

1. What are "Schottky defects" in crystalline solids?
2. What are "Frenkel defects" in crystalline solids?
3. How do you distinguish between crystal lattice and unit cell?
4. What are f - centers?
5. What is "coordination number"? Give this number in a (a) cubic close - pack structure (b) B.C.C. structure.
6. What is a) Ferromagnetism b) Antiferromagnetism?
7. What is a) Paramagnetism b) Ferrimagnetism?
8. What type of semi conductor is formed when
a) "Ge" doped with "In" b) "Si" doped with "B"

2. SOLUTIONS (6 Marks)

Short Answer type Questions

1. State Rault's law. Calculate the vapour pressure of a solution containing 9 g of glucose in 162 g of water at 293 K. The vapour pressure of water at 293 K is 17.535 mm Hg.
2. Calculate a) Molality b) Molarity and c) Mole fraction of KI if the density of 20% (mass/ mass) aqueous KI is 1.202 gm L^{-1} .
3. State Henry's law. Calculate Henry's law constant if the solubility of H_2S in water at STP is 0.195 m.
4. Define mole fraction. Calculate the mole fraction of H_2SO_4 in a solution containing 98% H_2SO_4 by mass.

Senior Inter Chemistry Important Questions Chapterwise

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