

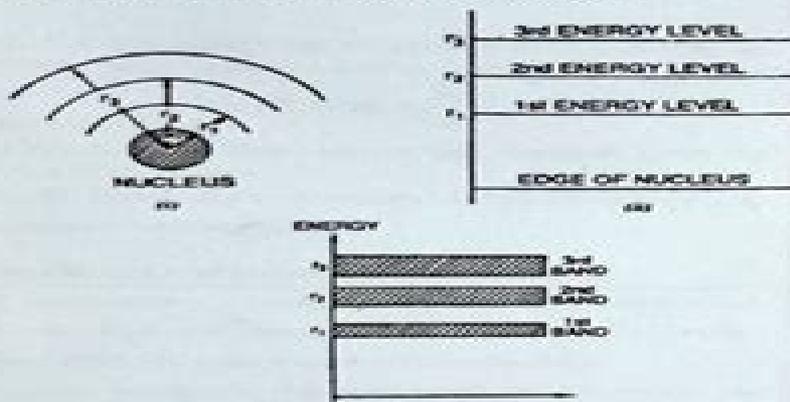


P.K.M PHYSICS

SEMICONDUCTOR DEVICES

Energy Levels

Each orbit has fixed amount of energy associated with it. The electron moving in a particular orbit possess the energy of that orbit. The larger the orbit, the greater is its energy and higher is the energy level. The first orbit represents the first energy level, the second orbit indicates the second energy level and so on.



Energy Bands

In case of a single isolated atom, the electrons in any orbit possess definite energy. When atoms bound together form a solid, each atom is influenced by presence of neighbouring atoms. So that the electron in any orbit of such an atom can have a range of energies rather than a single energy. This is known as energy band.

The range of energies possessed by an electron in a solid is known as energy band.

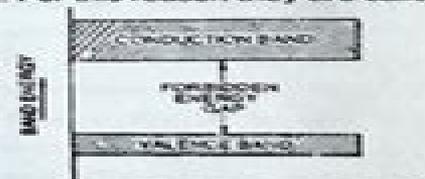
Important Energy Bands in solids

(1) **Valence Band** - The range of energies (i.e., band) possessed by valence electrons is known as valence band. The electrons in the outer most orbit of an atom

are known as valence electrons. Valence band has the electrons of highest energy. This band may be completely or partially filled. This band is never empty. For instance in case of inert gases, the valence band is full where as for other materials, it is only partially filled. The partially filled band can accommodate more electrons.

- (i) This is the band of maximum energy.
- (ii) Electrons are not capable of gaining energy from external electric field.
- (iii) The electrons in this band do not contribute to the electric current.
- (iv) The highest energy level which can be occupied by an electron in valence band at 0 K is called fermi level.

(2) **Conduction Band** - In certain materials the valence electrons are loosely attached to the nucleus. Even at ordinary temp, some of the valence electrons may get detached to become free electrons. It is these free electrons which are responsible for the conduction of current in a conductor. For this reason they are called conduction electrons.



The range of energies (i.e., band) possessed by conduction electrons is known as conduction band.

- (i) It is also called empty band of minimum energy.
- (ii) This band is either empty or partially filled by the electrons at room temperature.
- (iii) In this band the electrons can gain energy from external electric field and contribute to the electric current.
- (iv) The electrons in the conduction band are called the free electrons. They are able to move anywhere within the volume of the solid.

Semi Conductor Technical Manual

N Colangelo



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