

Name: _____

UNIT 4: Chemical Reactions Note Taking Document

Guidelines for making and writing observations:

1. Describe as many properties as you can.
2. Record detailed information that might help a chemist identify the substance.
3. Use descriptive words (such as thick, pink liquid) rather than imprecise language. For example, **DO NOT SAY**, "light-colored stuff."
4. Avoid opinions (such as looks gross)

Vocabulary:

- _____: something that can be _____ about a substance such as its color, smell, or boiling point.

1. Chemists rely on some properties that can only be observed under _____ such as _____ and _____ points.
2. Different substances have _____ melting and boiling points. Therefore, it is a _____ that could be used to help _____ a substance.
3. _____ is the temperature at which a solid will begin to melt and turn into a liquid.
4. _____ is the temperature at which a liquid will begin to boil and turn into vapor.

PHYSICAL CHANGE

1. Physical changes may change the state of matter but the substance _____ change.
 - Atoms do _____ change.
 - No new _____ form.
 - The state of matter or shape may _____
 - Can be _____

CHEMICAL CHANGE

1. Chemical changes create an entirely _____ substance. Referred to as _____
 - Brings about something _____
 - The chemical bonds of the molecules _____ and they reform in a different manner.
 - Chemical changes _____ be undone.

Key Concept

1. Different substances have _____ properties.

Note Taking Study Guide Chemical Reactions

Ingo Waldemar Dagobert Hackh



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Science Indiana Standards Manager Grade 6 McDougal Littell, 2004 *Chemical Reactions and Their Equations* Ingo Waldemar Dagobert Hackh, 1921 **Chemistry, Study Guide** Bernice G. Segal, Peter S. Shenkin, 1989-02-14 This Second Edition of the first year chemistry text known for its clarity of exposition and its large number of illustrative worked problems contains a more rigorous treatment of electrochemistry chemical equilibrium and thermochemistry Worked examples now number over 300 and exercises over 1460

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Introduction to Chemistry, Study Guide T. R. Dickson, 1999-07-07 Teaches chemistry by offering a dynamic provocative and relevant view of the topic and its importance to society and our daily lives Three themes are stressed throughout the text developing chemical thinking and chemical vision and refining problem solving skills Many chapters in this edition has been rewritten and rearranged to vitalize the topics and to include interesting examples analogies and images

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