

# Mole Lab

## Introduction to The Mole Concept



### Introduction

Although technically not a laboratory experiment, this activity certainly helps to drive home the main idea behind the mole concept—that chemists can count out infinitesimally small particles by weighing.

### Concepts

- Avogadro's number
- Chemical formulas
- Molar mass or molecular weight

### Materials (for each lab station, two or three lab pairs can work at each station)

- Aluminum foil
- Ammonium sulfate,  $(\text{NH}_4)_2\text{SO}_4$ , 100 g
- Sodium chloride,  $\text{NaCl}$ , 500 g
- Water, distilled, 300 mL
- Balance, electronic, 0.01-g precision (if only triple beam balances are available, it is best if there is one for each lab pair)
- Beakers, 600-mL, or cut off 2-L soda bottles, 3 labeled
- Cups, blue, clear, and red plastic, 2–3 of each
- Pipets, Beral-type, 2–3
- Spoons, plastic, 4–6

### Materials (for the instructor's station, preferably near the center of the room)

- Copper sheet, 12 pieces\*
  - Balance, electronic, 0.01-g precision
  - Cups, blue, clear, and red plastic, 1 each
  - Scoring Standard Table
- \*See the Preparation section below for size of copper pieces.*

### Safety Precautions

*Ammonium sulfate is slightly toxic by ingestion. Wear chemical splash goggles, chemical-resistant gloves, and a chemical-resistant apron. Wash hands thoroughly with soap and water before leaving the laboratory. Follow all laboratory safety guidelines. Please review current Material Safety Data Sheets for additional safety, handling, and disposal information.*

### Preparation

1. Label three 600-mL beakers “NaCl,” “water,” and “ammonium sulfate” and add to each the corresponding substance.  
*Note:* Do not write the chemical formula for ammonium sulfate on the beaker; students must determine this.
2. Place 2–3 spoons in each beaker containing a solid substance.
3. Place 2–3 pipets in the beaker of water.
4. Cut twelve pieces of copper sheet according to Table 1 and label with the corresponding numbers 1–12. *Note:* The copper pieces can have any weights you want them to. Using the weights given in the table allows you to use the Scoring Standard table as is. If you use other weights, just modify column C in the table accordingly, as shown in Figure 1 on page 2.

Cu Piece	Mass (g)
#1	4.22
#2	4.78
#3	4.98
#4	5.13
#5	5.44
#6	5.97
#7	6.09
#8	6.55
#9	6.79
#10	7.13
#11	7.65
#12	7.98

# Mole Lab Student Guide

**Brendan G. Carr**



## **Mole Lab Student Guide:**

This book delves into Mole Lab Student Guide. Mole Lab Student Guide is a crucial topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Mole Lab Student Guide, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
  - Chapter 1: Introduction to Mole Lab Student Guide
  - Chapter 2: Essential Elements of Mole Lab Student Guide
  - Chapter 3: Mole Lab Student Guide in Everyday Life
  - Chapter 4: Mole Lab Student Guide in Specific Contexts
  - Chapter 5: Conclusion
2. In chapter 1, this book will provide an overview of Mole Lab Student Guide. The first chapter will explore what Mole Lab Student Guide is, why Mole Lab Student Guide is vital, and how to effectively learn about Mole Lab Student Guide.
3. In chapter 2, this book will delve into the foundational concepts of Mole Lab Student Guide. This chapter will elucidate the essential principles that need to be understood to grasp Mole Lab Student Guide in its entirety.
4. In chapter 3, this book will examine the practical applications of Mole Lab Student Guide in daily life. This chapter will showcase real-world examples of how Mole Lab Student Guide can be effectively utilized in everyday scenarios.
5. In chapter 4, this book will scrutinize the relevance of Mole Lab Student Guide in specific contexts. This chapter will explore how Mole Lab Student Guide is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Mole Lab Student Guide. The final chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Mole Lab Student Guide.

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