

THE FOUR GOSPELS

The life and teachings of Jesus Christ from four perspectives.

Gospel

From *euangelion* (Greek), which means good news. This refers to Jesus' message, the good news of the Kingdom of God.

Why four Gospels?

- Different original audiences and different purposes.
- Four perspectives for a fuller picture.
- Repetition for emphasis.

MATTHEW

Author: tax collector called to become an apostle.

Original audience: Jewish readers.

Emphasis: on the fact that Jesus is the Messiah prophesied in the Old Testament.

Interesting fact: Matthew alone uses the term *kingdom of heaven* (equivalent to Kingdom of God).

LUKE

Author: the beloved physician and companion of Paul.

Original audience: Theophilus and gentile readers.

Emphasis: on Jesus' humanity and His kindness to the weak, suffering and outcasts.

Interesting fact: Luke alone tells of Jesus' experience at the temple at age 12.

Some things covered in all four Gospels:

- Ministry of John the Baptist.
- Feeding of the 5,000.
- Betrayal by Judas.
- Denial by Peter.
- Trial and crucifixion of Christ.
- Resurrection of Christ.
- Promise of His second coming.

MARK

Author: served with Paul, Barnabas and Peter.

Emphasis: on Jesus' mighty and miraculous works that prove He is the Messiah.

Interesting fact: "Only four paragraphs . . . are unique to Mark" (*Eerdmans' Handbook to the Bible*, p. 499).

JOHN

Author: fisherman called to become an apostle.

Original audience: gentile Christians at the end of the first century.

Emphasis: on the last days and hours of Christ's life so "that you may believe that Jesus is the Christ" (John 20:31).

Interesting fact: John alone gives Jesus' great "I am" statements and describes Jesus' washing of the apostles' feet on Passover.

Exposition Of The Four Gospels

R Barnett



Exposition Of The Four Gospels:

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 $2 \text{ H}_2(\text{g}) + \text{C}_2\text{H}_6(\text{g}) \rightarrow \Delta h = 94.5 \text{ kJ}$
 $\text{H}_2\text{O}(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \rightarrow \Delta h = 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) + \frac{7}{2} \text{O}_2(\text{g}) \rightarrow \text{CO}_2(\text{g}) + 3 \text{ H}_2\text{O}(\text{g})$ $\Delta h = 283 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g})$

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web thus for reaction 9 2 in which 1 mole of H_2O and 1 mole of OH^- combine to form 2 moles of H_2O $\Delta H = -55.8 \text{ kJ}$ as shown note that the enthalpy change $\Delta H_{\text{products}} - \Delta H_{\text{reactants}}$ is positive if heat is absorbed that is if $\Delta H_{\text{products}} > \Delta H_{\text{reactants}}$

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