

## Organic Chemistry 11th Edition Solomons Test Bank

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Package title: Solomons Test Bank  
Course Title: Solomons 11e  
Chapter Number: 2

Question type: Multiple choice

1) Which of the following compounds contains polar covalent bonds?

- A)  $\text{CS}_2$
- B)  $\text{LiF}$
- C)  $\text{F}_2$
- D)  $\text{CH}_3\text{F}$
- E) None of these choices.

Answer: D

Topic: Polar Covalent Bonds  
Section Reference 1: 2.2  
Difficulty: Easy

2) Which molecule does not have a dipole moment?

- A)  A benzene ring with two fluorine atoms at adjacent (ortho) positions.
- B)  A benzene ring with two fluorine atoms at meta positions.
- C)  A benzene ring with two fluorine atoms at para positions.
- D)  A benzene ring with fluorine atoms at the 1, 2, 4, and 5 positions.
- E) None of these choices.

Answer: D

Topic: Molecular geometry, polarity  
Section Reference 1: 2.3

Difficulty: Easy

3) Of the following solvents which one does *not* have a zero dipole moment?

- A) Pentane
- B) Cyclohexane
- C) Diethyl ether

# Organic Chemistry Solomons 11 Solution

**Joacim Rocklöv**



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