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web balance the following equations and indicate the type of reaction taking place 1 $\text{NaBr} + \text{H}_3\text{PO}_4 \rightarrow \text{HBr} + \text{Na}_3\text{PO}_4$ type of reaction 2 $\text{Ca(OH)}_2 + \text{Al}_2\text{SO}_4 \rightarrow 3 \text{CaSO}_4 + \text{Al(OH)}_3$ type of reaction 3 $\text{Mg} + \text{Fe}_2\text{O}_3 \rightarrow \text{Fe} + \text{MgO}$ type of reaction 4 $\text{C}_2\text{H}_4 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ type of reaction 5 $\text{PbSO}_4 + \text{PbSO}_3 + \text{O}_2$ type of reaction 6 $\text{NH}_3 + 12$