



# Restaurant Start Up Quick Guide

# Restaurant Start Up Quick Guide

**Scott C. Dulebohn**



## **Restaurant Start Up Quick Guide:**

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web radioactive decay aqa with the wrong number of neutrons nuclei can fall apart a nucleus will regain stability by emitting alpha or beta particles and then cool down by emitting gamma

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web the first two equations have been balanced for you alpha particle  $\alpha$  2 4 beta particle  $\beta$  gamma ray  $\gamma$  proton p neutron n 0 positron  $\beta^+$  0 1 30 1 65 1 2 63 2 15 0 30 0 65 30 9 29 2 2 12 6 15 32 30

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web we can write a balanced nuclear equation for this reaction as let s do a practice example identify the product x by writing a balanced nuclear equation  $^{137}_{55}\text{Cs} \rightarrow ^{137}_{56}\text{Ba} + x$  because the mass number is the same on both sides of the

equation x must have a

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naoh na<sub>3</sub> aso<sub>3</sub> h

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web identify the missing particle in the following nuclear reaction  ${}^1_0\text{n} + {}^{235}_{92}\text{u} \rightarrow {}^2_{10}\text{n} + {}^{137}_{52}\text{te} + 4\text{}^2_2\text{he} + 97\text{}^1_0\text{n}$   ${}^{97}_{40}\text{zr}$

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web the coefficient in a balanced equation is an idea the concept of telling the chemist that if the atoms molecules and  
compounds are balanced there are balanced amounts of the atoms molecules and compounds on the other side in the product

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212 beryllium 8 and a positron are produced by the decay of an unstable nucleus neptunium 239 forms from the reaction of  
uranium 238 with a neutron and then spontaneously converts into plutonium 239 strontium 90 decays into

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reaction  ${}^6_5\text{i} \rightarrow {}^9_4\text{be} + \text{e}^+$  express your answer as an integer hint 1 how to approach the problem use rule 2 the total of the subscripts  
atomic numbers mathematically this can be written as answer

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atomic mass of 80 amu an element emits two neutrons while decaying into two metals each of which can be extracted and  
converted to chlorides with the formula  $\text{mcl}_2$

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preview multiple choice 5 minutes 1 pt complete the nuclear equation and determine the type of decay that is occurring in  
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