

RE: _____
Patient/Client: _____
Date of Accident: _____
Our File No: _____

Dear _____:

I represent _____ in her claim for personal injuries arising out of an accident which occurred on the _____ day of _____, 20____, at _____ in _____, _____.

I am enclosing a medical authorization form signed by my client and request that you forward to me a comprehensive but concise narrative medical report that encompasses the following:

1. A description of the accident as related by my client combined with an opinion as to the reasonableness of the accident being the causative agent of the subsequent physical complaints;
2. A description of the Plaintiff's specific subject complaints;
3. The pertinent past medical history taken from my client and any probable or feasible association with the current complaints, and if applicable, address the matter of traumatic aggravation of any pre-existing conditions;
4. The pertinent subjective and objective physical findings, along with an opinion as to their significance, appropriateness, and relevancy;
5. The significance of the results of any diagnostic tests performed by you combined with an assessment of the consistency of these tests with the medical history and physical findings;

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WJ Hussar



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ions that remain essentially unchanged during a reaction are called spectator ions this means you can ignore them when you write the ionic equation you only need to model how the solid

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highlights learning objectives by the end of this section you will be able to derive chemical equations from narrative descriptions of chemical reactions write and balance chemical

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the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction notice that in writing the net ionic

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what remains when the spectator ions are removed is called the net ionic equation which represents the actual chemical change occurring between the ionic compounds cl aq

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this net ionic equation indicates that solid silver chloride may be produced from dissolved chloride and silver i ions regardless of the source of these ions these molecular and

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write the overall chemical equation the complete ionic equation and the net ionic equation for the reaction of aqueous silver fluoride with aqueous sodium phosphate to give solid silver

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the complete ionic equation for this reaction is as follows equation 4 13 2aq aq 2f aq 2nh₄ aq cr₂o₇²⁻ aq ag₂cr₂o₇ s 2nh₄ aq 2f aq because two nh₄ aq

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