

Chapter 12: Crystallography - Nuclear Chemistry

12. Nuclear Reactions

- chemical reactions involve e^- transfer, nuclear unaltered
- nuclear reactions involve Δ in the number of different elements
- called α transformations

A. Stability of Nuclei

- depends on # of protons (positive e^-) repel each other
- ratio of n to p increases for heavy particles, greater stability
- elements $Z > 82$ and above: all are unstable (radioactive)

B. Notation Symbols

$${}^A_Z X \rightarrow \text{where } A = \text{mass number of } X$$

C. Types of Radioactive Decay

1. Alpha decay - spontaneous emission of helium nuclei

(Original nucleus decays \rightarrow new nucleus)

2. Radioactive decay reactions \rightarrow 4 types

- a) Alpha decay ${}^A_Z X \rightarrow {}^{A-4}_{Z-2} Y + {}^4_2\text{He}$ \rightarrow particle is $2p$ and $2n$
 \rightarrow Δ Z change
 \rightarrow new penetrating power

- b) Beta decay ${}^A_Z X \rightarrow {}^A_{Z+1} Y + e^-$ \rightarrow neutrally charged

\rightarrow beta particle is an electron, Δ Z ± 1 change
 \rightarrow beta decay "relating" $n \rightarrow p$ process, beta and beta-
transformation are the same process

Neutrinos \rightarrow \rightarrow \rightarrow (high speed particles)
(gamma rays)

$${}^A_Z X \rightarrow {}^A_{Z+1} Y + e^- + \bar{\nu}$$

$$\text{where } \bar{\nu} = \text{anti-neutrino}$$

\rightarrow moderate penetrating (through clothing, through skin)

- c) Gamma decay ${}^A_Z X \rightarrow {}^A_Z Y + \gamma$

Physical Nuclear Chemistry Topic 12 Answers

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