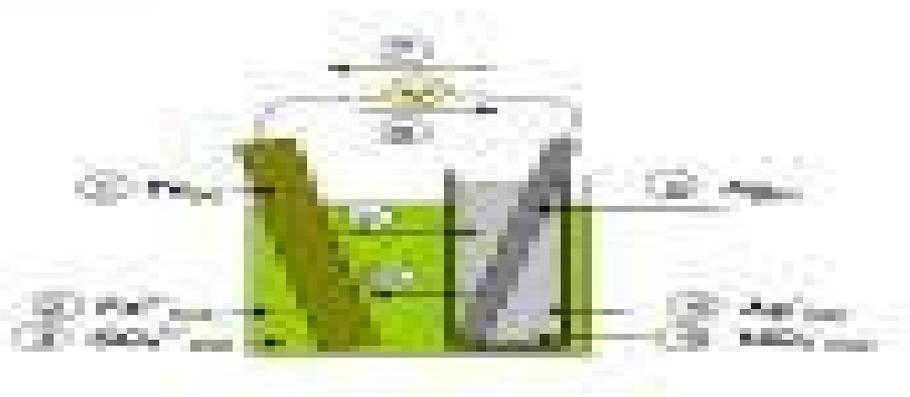


# Voltaic Cells

Label the parts for the following cell:



By connecting to the voltmeter, describe the components and their function. (10 marks)

- a. Zn electrode
- b. ZnSO<sub>4</sub> solution
- c. Cu electrode
- d. CuSO<sub>4</sub> solution
- e. Salt bridge (connects the two solutions)
- f. External circuit (connects the two electrodes)
- g. Voltmeter (measures the potential difference)
- h. Zn
- i. Both Zn and Cu are oxidized and reduced in the cell, as shown by the reaction:  $Zn + Cu^{2+} \rightarrow Zn^{2+} + Cu$

.....

# Voltaic Cells Study Guide

**J Ma**



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